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Question 2 1 pts What is the mass % (m/m) of a solution containing 15.5 g of sodium hypochlorite and 100.0 g of water. The solution is % NaClO. (Your answer should have 3 significant figures.)
Question 3 1 pts What is the mole fraction of sodium hypochlorite (x_{NaClO}) in a solution containing 15.5 g of sodium

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hypochlorite and 100.0 g of water.

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X Your answer: For webquest or practice, print a copy of this quiz at the Chemistry: Solutions webquest print page. About this quiz: All the questions on this quiz are based on information that can be found at Chemistry: Solutions .

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Because the question involves mass, we will need to know the molar mass of $C_6H_{12}O_6$. Using a periodic table we find the molar mass of $C_6H_{12}O_6$ to be $180.1 \text{ g}\cdot\text{mol}^{-1}$.

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